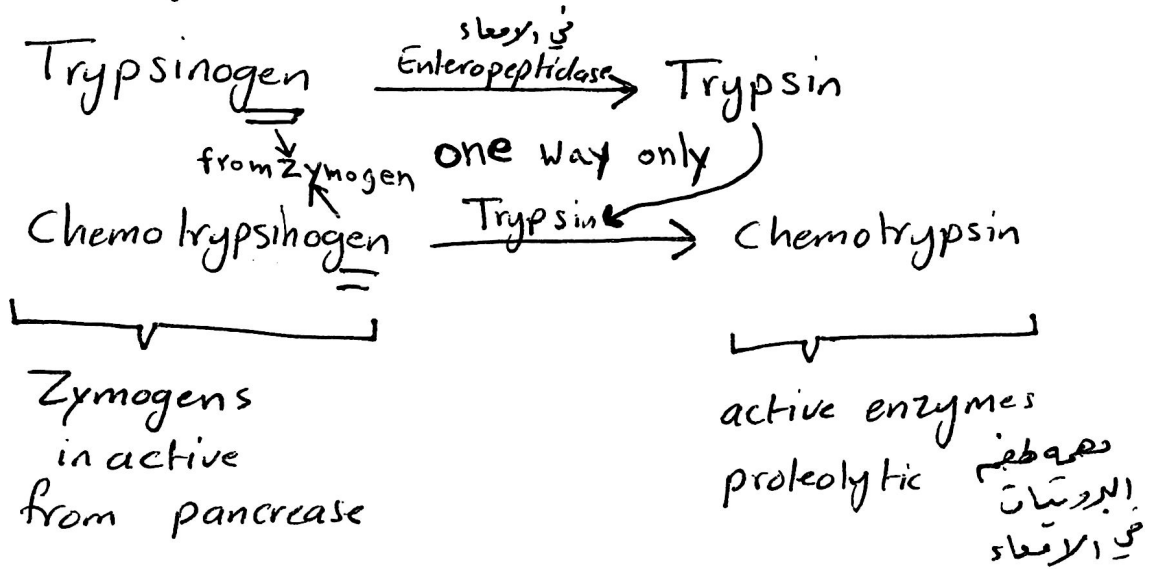


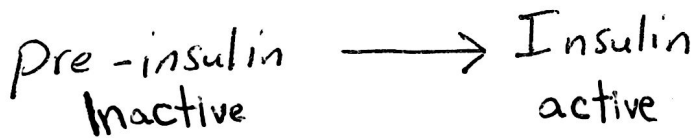
السلطة الأولية للزيم
Zymogens: inactive precursor of the enzyme

recomes → activated by breaking peptide (covalent) bonds
(Irreversible / covalent not allosteric)

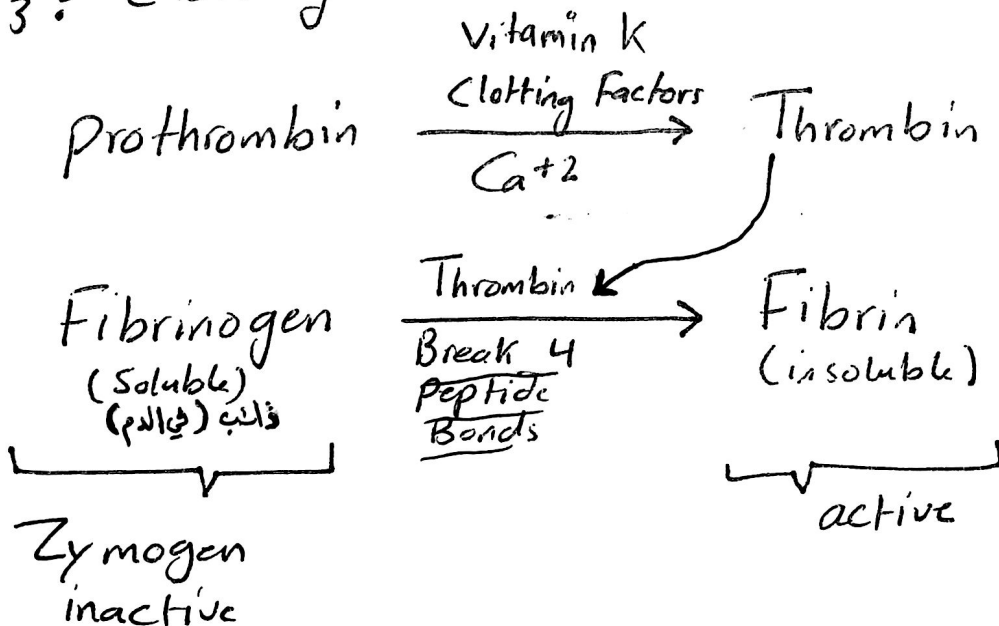
Ex₁: Digestive Enzymes



Ex₂: Hormons

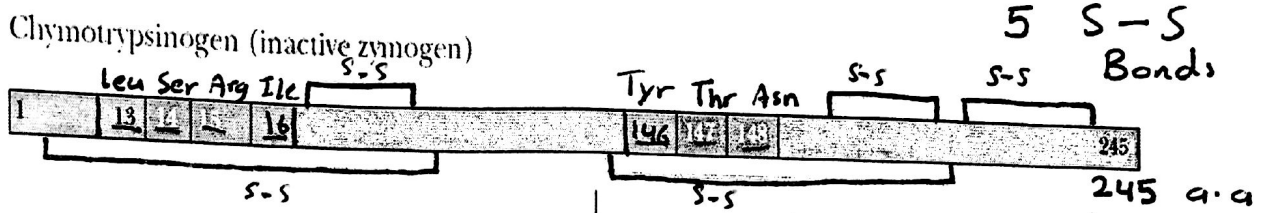


Ex₃: Clotting Factors

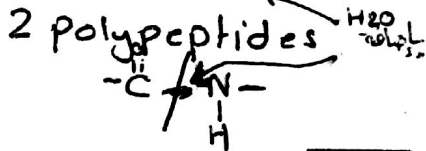
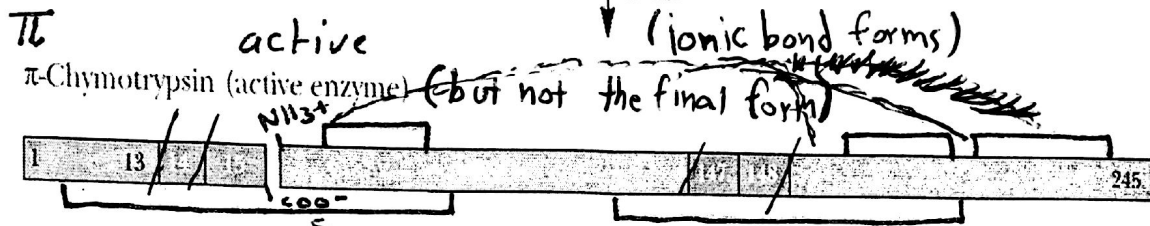


* Hemophilia: Genetic disease affecting clotting

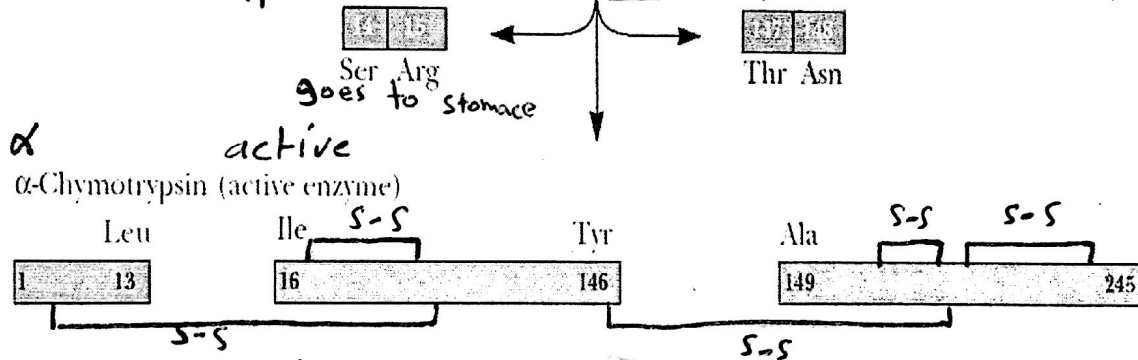
activation of chymotrypsinogen



Cleavage at Arg¹⁵ by trypsin



Self-digestion at Leu¹³, Tyr¹⁴⁶, and Asn¹⁴⁸ by π -chymotrypsin



Final Form

3 Polypeptides connected by S-S Bonds

Change 1^o → Change 3^o
Become active.

* From X-Ray Crystallography

The amino-end of Isoleucine 16 after First NH₃⁺

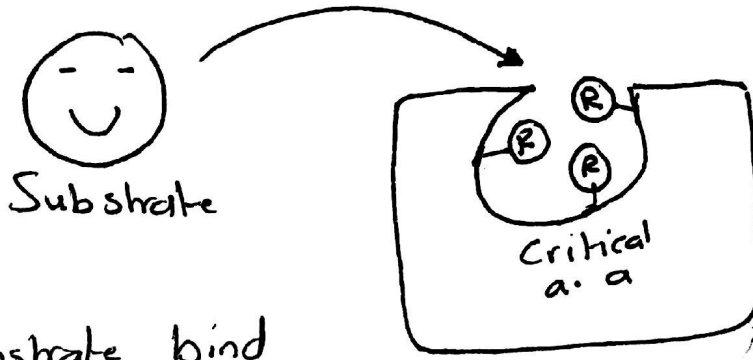
Cleavage make ionic bond with Asp 194

Necessary for activity

(Chymotrypsinogen Lack this ionic bond)

Active Site

الموقع
النشط



* the substrate bind
to R-groups in the active site
these R-groups are essential and play
Catalytic role

* the most common R-groups that play
Catalytic role in Enzymes

- 1 - Imidazole of Histidine
- 2 - OH of Serine
- 3 - COO^- of Aspartic acid and Glutamic acid
- 4 - SH of Cystine
- 5 - amino group of Lysine (just understand them)
- 6 - phenol group of Tyrosine
- 7 - α -amino-end
 α -Carboxyl end

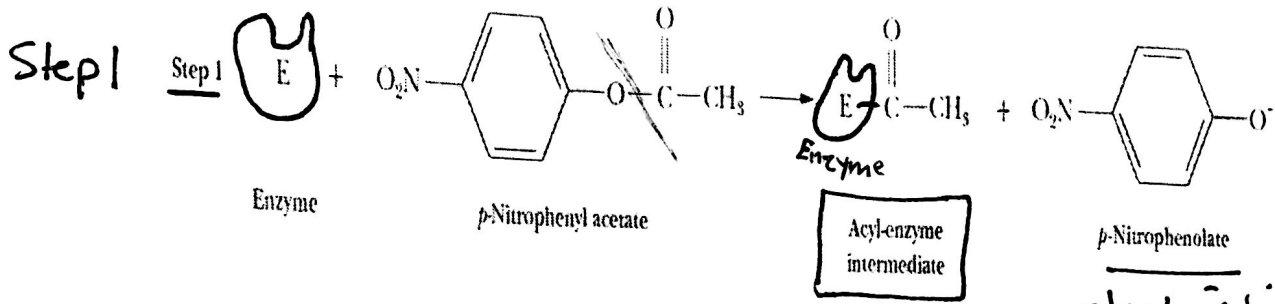
Which group of amino-acids are likely to be found in the active site of an enzyme?

- a. leucine, lysine, alanine nonpolar
- b. Cystein, isoleucine, phenylalanine nonpolar
- c. tyrosine, threonine, leucine nonpolar
- d. Serine, histidine, aspartate

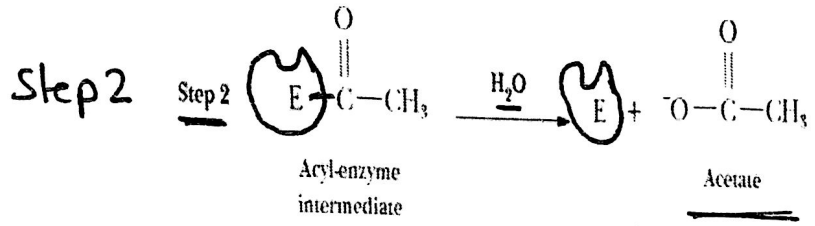
Q: The ^(R-Groups) amino-acids in the active site can be involved in all of these processes, except :-

- a. Binding of the substrate
- b. Becoming part of the product of the reaction
- c. The actual chemical mechanism for the reaction
- d. Binding of some necessary cofactor for the reaction
- e. all of these can be functions of the amino-acids in the active site

Chymotrypsin works in 2 steps

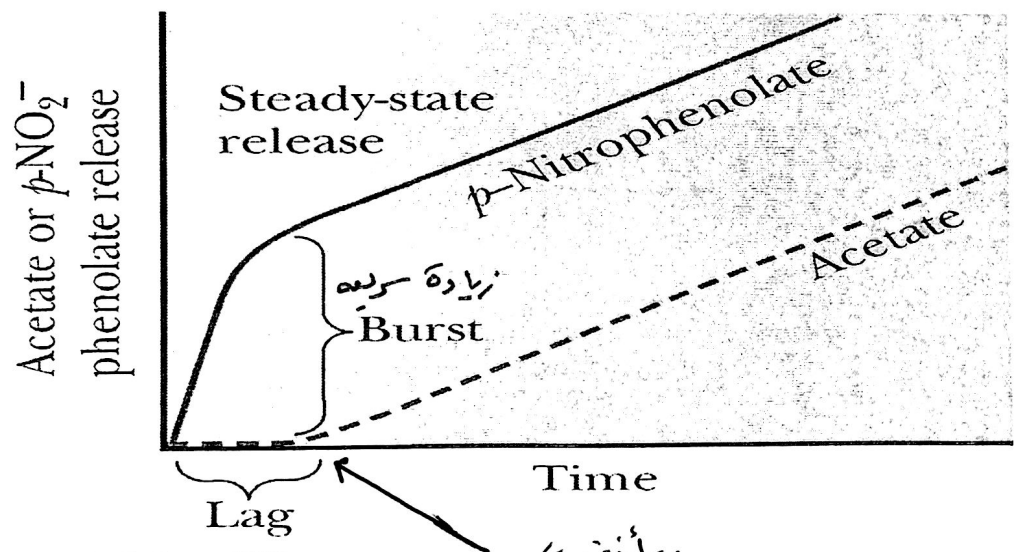


الناتج الاول
يكرر بسرعة
Burst



الناتج الثاني
Lag

© 2006 Brooks/Cole - Thomson



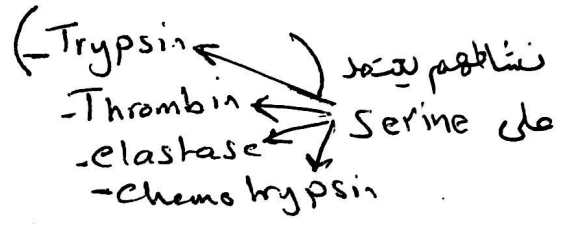
يتأخر في تكون
تركيزه لأنه
يظهر في الخطوة الثانية
فترة تكون

© 2006 Brooks/Cole - Thomson

Active Site of Chemotrypsin

it was found that 2 amino-acids in the active site of chemotrypsin are essential (critical for) activity

Serine 195 → so it belongs to Serine-protease Family



Histidine 57

But how do we know that?

Labeling:- binding a substance by covalent bond to a specific side chain (R)
Covalent modification

Serine 195 labeled by DIPF → Chemotrypsin becomes inactive.
when added:
R groups = المجموعات الجانبية

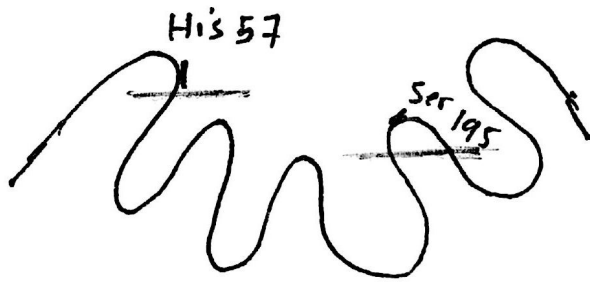
Histidine 57 labeled by TPCK → chemotrypsin becomes inactive

Q: Inhibitors (ex. DIPF, TPCK) which bind covalently to specific amino-acids are useful in determining which amino-acids are in the active site of the enzyme

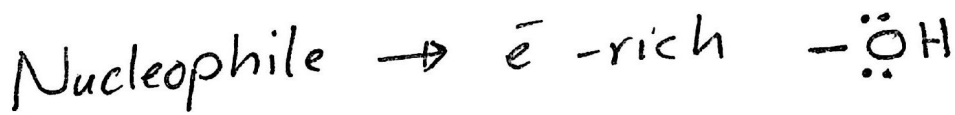
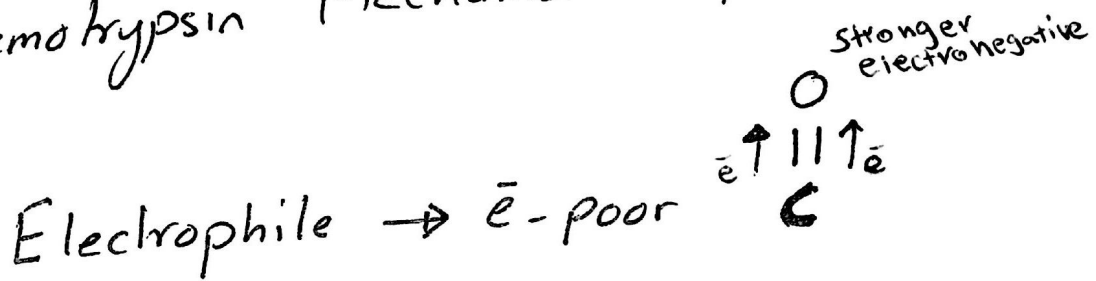
- A. True
- B. False

Site !!

X-ray Crystallography Show that folding of chymotrypsin is in an anti-parallel β -sheets positioned them in the active site pocket



Chemotrypsin Mechanism of action

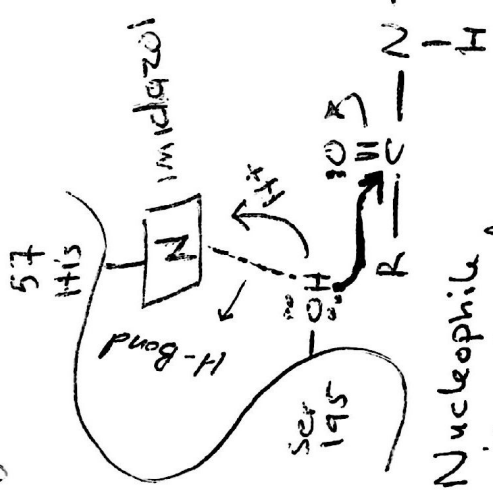


* Usually Nucleophile attack and bind to Electrophile

Chemotrypsin Mechanism of action

Stage 1

Tetrahedral formation

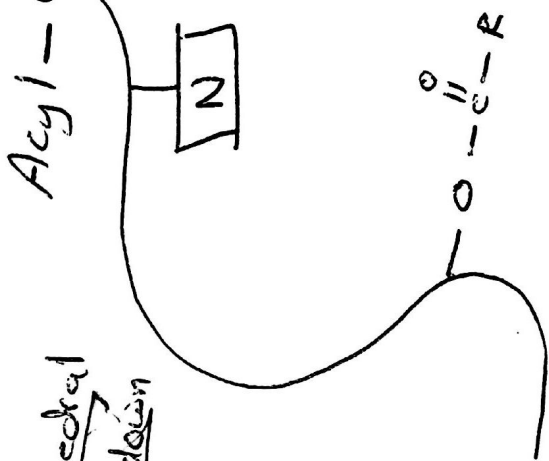


* Nucleophile is OH of Serine

* His (Imidazole) act as General Base Catalysis

Tetrahedral intermediate

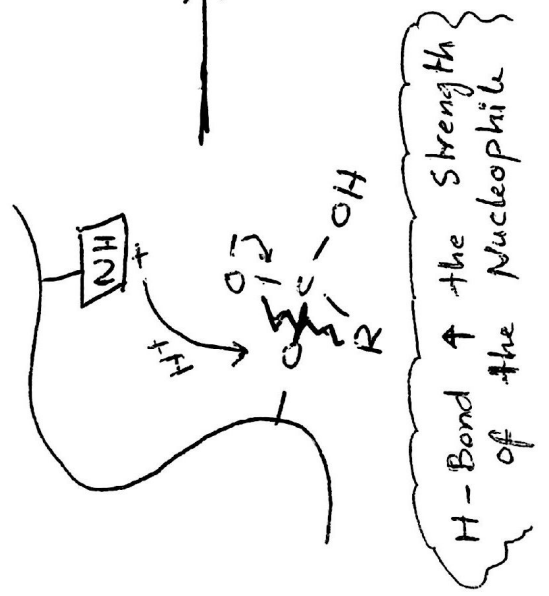
Tetrahedral Breakdown



+ ~~H2O~~
H2N-R
"Burst"

* His (Imidazole) act as General acid Catalysis

Stage 2



* Nucleophile is H2O Double bond

R-C(=O)-OH
"Lag"

Q: The initial bond formation in the covalent intermediate in the reaction catalyzed by chymotrypsin is between:-

- a. Serine and the carbonyl carbon in the peptide backbone
- b. Serine and the nitrogen in the peptide backbone
- c. histidine and the carbonyl carbon in the peptide backbone
- d. histidine and the nitrogen in the peptide backbone

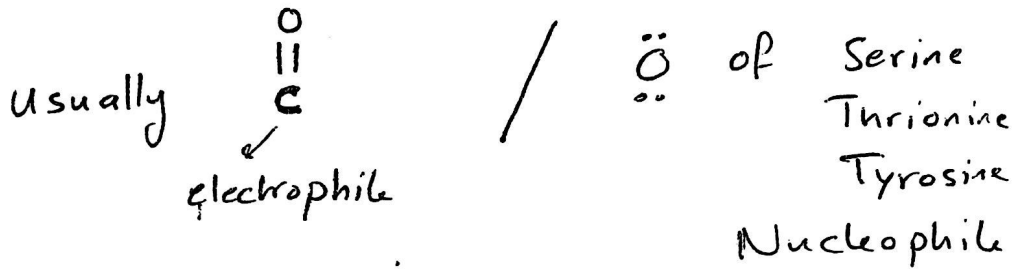
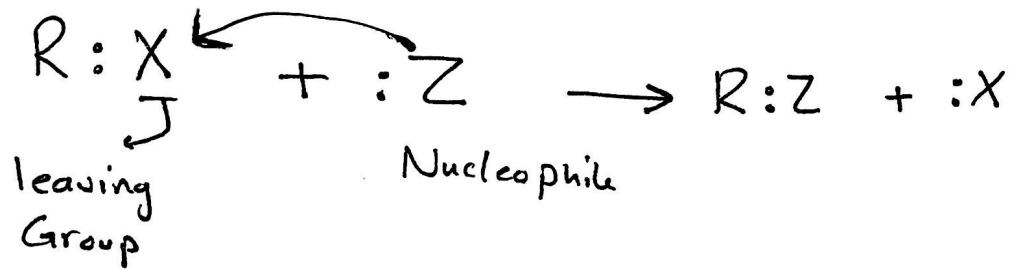
Q: An important step in elucidating the behavior of an enzyme is:-

- a. Obtaining a crystalline sample of the enzyme
- b. insuring that metal ions are always excluded from the enzyme sample
- c. determining the active site residues
- d. none - of - these.

Common Reaction in Enzyme mechanism

I Nucleophilic Substitution Reaction (S_N1, S_N2)

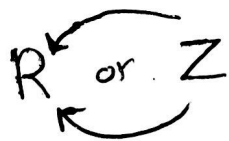
المركب,
المادة



* IF the rate of the reaction depends on [R:X] Only, then the reaction is

S_N1 → First order

Here :X leave then :Z attack quickly
 rate limiting step



↳ Can attack from any direction

2 stereoisomers produced

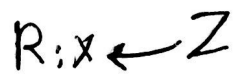
S_N1 Not stereospecific
 (does not give 1 stereoisomer)

* IF the reaction rate depends on both [R:X] and [Z] then the reaction is S_N2 → Second order

Here :Z attack while :X still bound

↳ attack from one direction
 produce one stereoisomer

S_N2 stereospecific



because $\overset{\text{O}}{\parallel}{\text{C}}$ return $\overset{\text{O}}{\parallel}{\text{C}}$ → Not chiral
 Achiral carbonyl No stereoisomers
 At the end of reaction

2 Acid-Base Catalysis

→ according to Bronsted-Lowry

acid → H^+ donor

Base → H^+ acceptor

General acid-Base Catalysis

IF the amino-acid act as H^+ donor → General acid Catalysis

IF the amino-acid act as H^+ acceptor → General Base Catalysis

Imidazol, OH, COOH, SH, NH, phenol

General acid - Base Catalysis

Lewis acid-Base Catalysis

according to Lewis

acid: \bar{e} -pair acceptor (\bar{e} -poor)

base: \bar{e} -pair donor (\bar{e} -rich)

Metal ions Mn^{+2} , Mg^{+2} , Zn^{+2} in Enzymes are Lewis acids

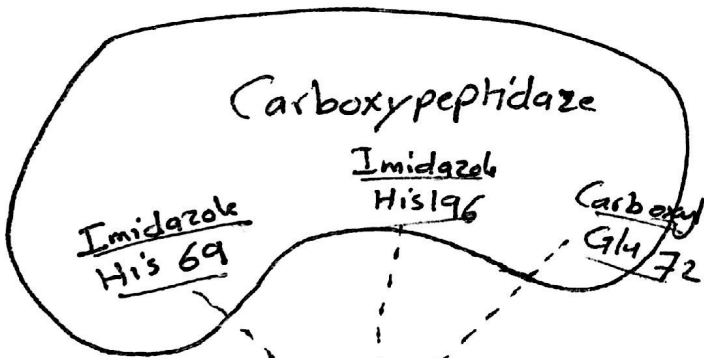
Ex:- Zn^{+2} in Carboxypeptidase

انزيم يهضم البروتينات

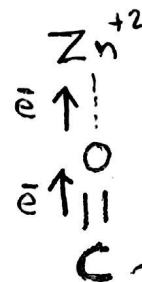
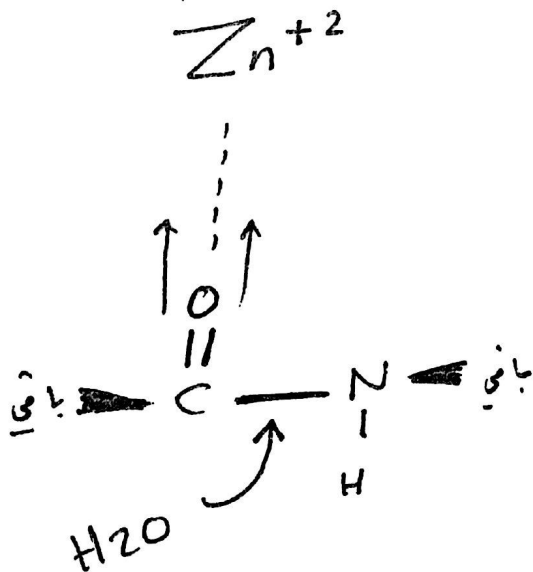
Hydrolyze

تبدأ من ناحية
Carboxyle

the C-terminal



coordination Bond



تصبح فقيرة
الالكترونات

معرضة اكثر
More Susceptible
for water attack

specificity of the active site

a. absolute specificity : One and only one
تعلق وظيفي
Substrate bind
One reaction
One product

→ Rigid active site

→ lock and key

b. Relative specificity : Similar Substances
تعلق وظيفي
Can bind

→ flexible active site

→ Induced Fit model

Steriospecific Enzymes:

Enzymes bind to specific Stereoisomer

either L or D,

Here the active site is chiral, 

and the active site must be

as the substrate not the mirror image

L binds with
L - L

D - D



Factors ^{العوامل المساعدة}

Cofactors are Non-protein substances, take part in enzymatic reaction and regenerated at the end of the reaction

* Cofactors may bind to the enzyme by covalent or non-covalent bond.

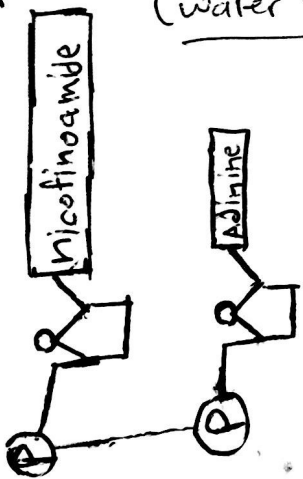
Ex:- Metal ions ; act as lewis acids

- Coenzymes → Organic Compounds, many of them are Vitamins or related to vitamins.

* B vitamins important in oxidation-reduction reaction

(water soluble)

or Group transfer reaction



Ex₁:- NAD^+ (DNA, RNA يكو)

Nicotinamide - Adenine - Dinucleotide

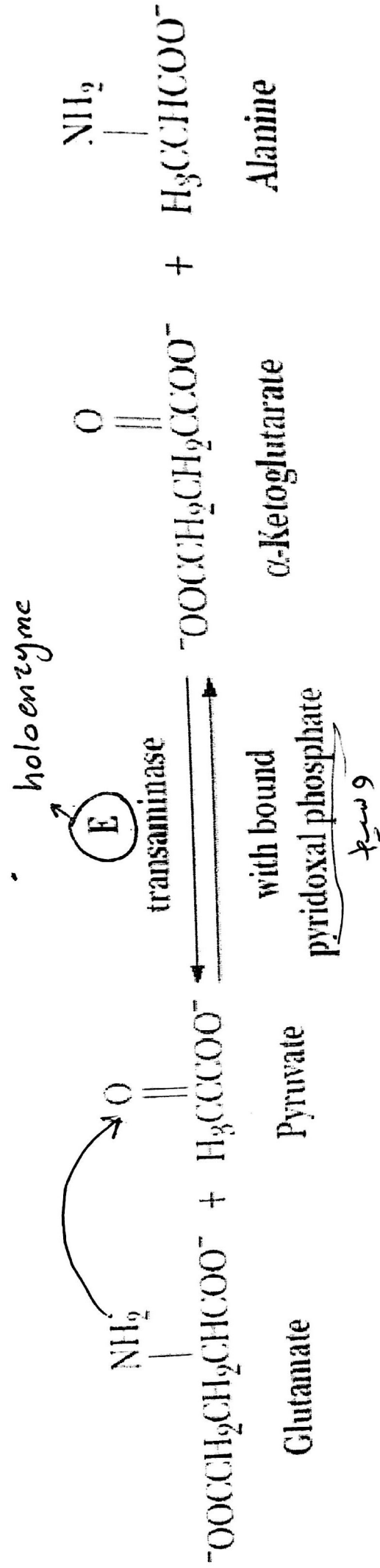
derived from Vitamin Niacin (Nicotinic acid)

Nicotinamide Ring is the active site of redox

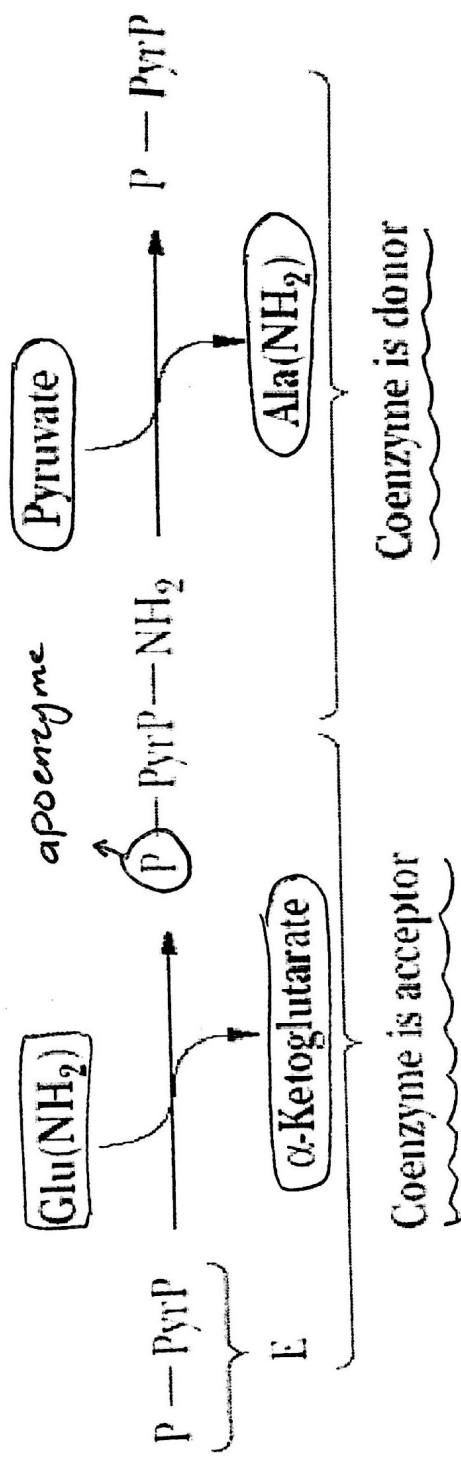
Ex₂: Vitamin B₆

1-Pyridoxal 2-Pyridoxamine 3-Pyridoxin

active forms → 1-Pyridoxal phosphate 2-pyridoxamine phosphate
transfer amino-group from one substance to another (transamination)



This amino (NH₂) group transfer reaction occurs in two stages:



© 2006 Brooks/Cole - Thomson

The role of pyridoxal phosphate as a coenzyme in a transamination reaction.
 PyrP is pyridoxal phosphate, P is the apoenzyme (the polypeptide chain alone), and E is the active holoenzyme (polypeptide plus coenzyme).

Table 7.1

Coenzymes, Their Reactions, and Their Vitamin Precursors

Coenzyme	Reaction Type	مشتقون Vitamin Precursor	See Section
Biotin	Carboxylation	Biotin	18.2, 21.6
Coenzyme A	Acyl transfer	Pantothenic acid	15.7, 19.3, 21.6
Flavin coenzymes (FAD, FMN)	Oxidation-reduction	Riboflavin (B ₂)	15.7, 19.3
Lipoic acid	Acyl transfer	—	19.3
Nicotinamide adenine coenzymes NAD ⁺ NADP ⁺	Oxidation-reduction	Niacin (B ₃)	15.7, 17.3, 19.3
Pyridoxal phosphate	Transamination	Pyridoxine (B ₆)	23.4
Tetrahydrofolic acid	Transfer of one-carbon units	Folic acid (B ₉)	23.4
Thiamine pyrophosphate (TPP)	Aldehyde transfer (Decarboxylation)	Thiamine (B ₁)	17.4, 18.4

wish you all Luck
 Dr. Tariq. Jibril
 0790 979 188